

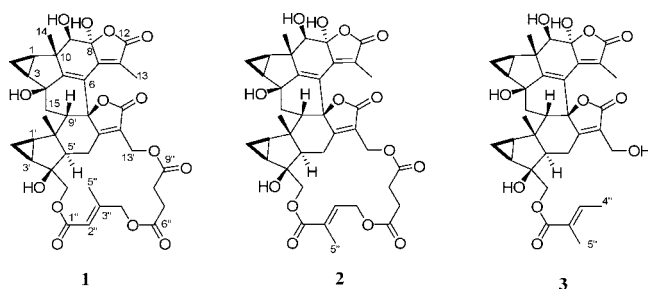
Mono- and Di-sesquiterpenoids from *Chloranthus spicatus*Yong-Jiang Xu,[†] Chun-Ping Tang,[†] Chang-Qiang Ke,[†] Ji-Bao Zhang,[†] Hans-Christoph Weiss,[‡] Ernst-Rudolf Gesing,[§] and Yang Ye^{*†}

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Three new dimeric sesquiterpenoids, chloramultilides B–D (**1–3**), along with 10 known sesquiterpenoids, were isolated from the whole plant of *Chloranthus spicatus*. Their structures were established by physical data (1D and 2D NMR, MS). The structure and absolute configuration of **1** was confirmed by X-ray crystallography. Compound **1** exhibited moderate *in vitro* antifungal activity.

Chloranthus spicatus (Thunb.) Makino (Chloranthaceae), distributed mainly in southern China, has long been regarded as a medicinal material to treat aches, trauma, and bleeding. Its roots, known as “zhulan-gen” in China, were used externally to cure carbuncle, furuncle, tumefaction, and ringworm.¹ So far, sesquiterpenoids^{2–4} and dimeric sesquiterpenoids^{5–11} were isolated from the genus *Chloranthus* as their major secondary metabolites. Diterpenoids were reported also from this genus.¹² Previous pharmacological studies revealed that sesquiterpenoid monomers exhibit antifungal activities,¹³ and sesquiterpenoid dimers show tumor growth inhibitory activities.¹⁴ Yang et al. reported that sesquiterpenoid dimers exhibit potent and selective inhibition on the delayed rectifier (I_K) K⁺ current.¹⁵ In our investigation on the whole plant of *C. spicatus*, three new dimeric sesquiterpenoids (**1–3**) were isolated and identified, along with 10 known compounds. The new structures were established on the basis of 1D and 2D NMR data, as well as other spectroscopic analyses. The structure and absolute configuration of **1** were confirmed by X-ray crystallography. The structures of known compounds were elucidated by comparison with reported data.

Structures of chloramultilides B–D (**1–3**).

Results and Discussion

Chloramultilide B (**1**) was obtained as an amorphous powder. The molecular formula was assigned as C₃₉H₄₂O₁₄ by HRESIMS. The IR spectrum displayed absorption bands for hydroxy (3450 cm⁻¹) and ester (1762 cm⁻¹) groups. The ¹³C NMR spectrum displayed 39 carbon resonances, which were ascribed to five carbonyl, eight olefinic, four methyl, nine methylene, seven methine, and six quaternary carbons (Table 2). The ¹H NMR spectrum exhibited characteristic resonances of one vinylic proton (δ_{H} 6.09,

br s) and four methyl singlets (δ_{H} 1.09, 1.15, 1.90, and 2.01). The above data revealed that **1** might be a sesquiterpenoid dimer. Analysis of the ¹H and ¹³C NMR spectra indicated that the NMR data of **1** (Tables 1 and 2) strongly resembled those of chloramultilide A.¹¹ One major difference between these two compounds was the chemical shift of C-8. In comparison with the corresponding resonance (δ_{C} 199.3) in chloramultilide A, C-8 in compound **1** shifted upfield to δ_{C} 104.9, which was indicative of a hemiacetal atom, not a ketone carbon. The C-8 was furthermore connected to C-12 via an oxygen atom to form a five-membered α,β -unsaturated lactone ring fused at C-7 and C-8 by the key HMBC correlations between the allylic methyl (δ_{H} 2.01) and C-6 (δ_{C} 122.5), C-7 (δ_{C} 155.0), C-8, C-11 (δ_{C} 122.0), and C-12 (δ_{C} 172.9). The other differences, in comparison with the corresponding resonances of chloramultilide A, were the upfield shift of the olefinic carbon C-2'' (δ_{C} 114.0) and downfield shift of the olefinic carbon C-3'' (δ_{C} 152.1), which might be elucidated by the location of Me-5'' at C-3'' instead of C-2''. Thus the planar structure of **1** was established.

The relative configuration of **1** was assigned by a ROESY experiment. The correlations of H-1/H-3, H-1/H-9, H-3/H-9, H-9/H-15 α , H-2 β /Me-14, H-1'/H-3', H-2' β /Me-14', H-1'/H-2' α , H-3'/H-2' α , and H-5'/H-15' α indicated the orientation of H-1 (α), H-3 (α), H-9 (α), Me-14 (β), H-1' (α), H-3' (α), H-5' (α), and Me-14' (β), respectively (Figure 1). Considering its biogenetic relationship, such elucidation was fully consistent with those of naturally occurring lindenane-type sesquiterpenoids.

An X-ray crystallographic diffraction experiment was carried out to confirm the absolute configuration of **1** (Figure 2). The crystal structure was determined with Cu K α radiation at 100 K with 99% coverage and an averaged redundancy of 8.9. Refinement of the Flack parameter gave a value of 0.01(15) for the absolute configuration depicted. Additionally, the absolute configuration was checked by a statistical analysis of the Bijvoet pairs implemented in Platon.¹⁶ Finally, a second crystal of the same batch was measured. In all cases the result of the assignment of the absolute configuration could be confirmed.

The formula for the contents of the asymmetric unit of the crystal structure is C₃₉H₄₂O₁₄ (for **1**), solvated with a molecule of dimethylformamide (C₃H₇NO) and half a molecule of water; water H atoms could not be located. Consequently, the formula of the asymmetric unit differs from the formula of the molecule. The given molecular weight of 815.82 belongs to the whole asymmetric unit.

Chloramultilide C (**2**) was obtained as an amorphous powder, whose positive ion ([M + Na]⁺ *m/z* 757.2443) in the HRESIMS indicated the same molecular formula as that of **1** (C₃₉H₄₂O₁₄). The IR absorption bands at 3428 and 1753 cm⁻¹ suggested the presence of hydroxy and ester groups. The ¹H and ¹³C NMR data of **2** were

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Table 1. ^1H NMR (300 MHz) Data of Compounds **1–3** (pyridine- d_5)

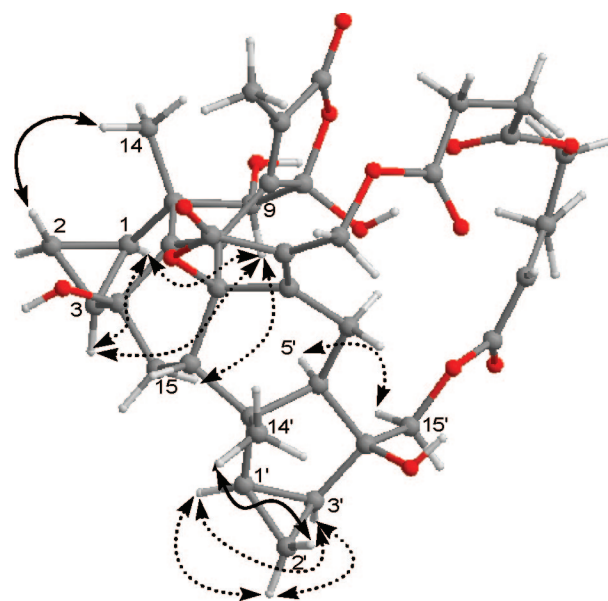
no.	1	2	3
1	2.21, m	2.20, m	1.81, m
2	0.98, m 1.45, m	0.95, m 1.42, m	0.79, m 1.03, m
3	2.05, m	2.00, m	1.79, m
9	4.48, s	4.48, s	3.67, s
13	2.01, s	2.02, s	1.52, s
14	1.09, s	1.09, s	0.80, s
15	2.20, m 3.03, m	2.10, m 3.04, m	1.82, m 2.67, m
1'	1.72, m	1.70, m	1.61, m
2'	0.63, m 1.60, m	0.63, m 1.62, m	0.65, m 1.19, m
3'	1.59, m	1.58, m	1.65, m
5'	2.81, m	2.90, m	2.39, m
6'	3.60, m 3.68, m	3.42, m	2.41, m 2.89, m
9'	3.25, m	3.20, m	2.68, m
13'	4.91, d (11.9) 5.32, d (11.9)	4.93, d (12.0) 5.30, d (12.0)	4.25, m
14'	1.15, s	1.13, s	0.96, s
15'	4.21, d (15.4) 5.10, d (15.4)	4.18, d (11.4) 5.20, d (11.4)	3.98, d (11.1) 4.02, d (11.1)
2''	6.09, br s		
3''		7.02, br s	6.97, m
4''	4.21, d (9.6) 4.81, d (9.6)	4.20, d (7.4) 4.82, d (7.4)	1.89, d*
5''	1.90, s	1.52, s	1.78, s
7''	2.59, m	2.60, m	
8''	2.61, m	2.70, m	

* Overlapped by other resonances.

Table 2. ^{13}C NMR (100 MHz) Data of Compounds **1–3** (pyridine- d_5)

no.	1	2	3
1	29.6, d	29.9, d	29.7, d
2	9.9, t	9.9, t	9.9, t
3	31.5, d	31.5, d	31.5, d
4	77.3, s	77.4, s	77.5, s
5	164.2, s	164.3, s	164.4, s
6	122.5, s	122.5, s	123.7, s
7	155.0, s	154.7, s	155.1, s
8	104.9, s	104.7, s	104.7, s
9	79.8, d	80.5, d	86.0, d
10	49.6, s	51.0, s	50.6, s
11	122.0, s	122.0, s	124.1, s
12	172.9, s	172.9, s	172.7, s
13	10.8, q	10.9, q	10.8, q
14	14.4, q	12.4, q	14.1, q
15	41.7, t	41.7, t	41.8, t
1'	27.5, d	27.3, d	27.5, d
2'	11.0, t	10.7, t	11.0, t
3'	29.3, d	29.6, d	29.8, d
4'	77.2, s	77.0, s	77.3, s
5'	56.1, d	55.9, d	52.8, d
6'	25.6, t	25.0, t	22.0, t
7'	177.3, s	176.8, s	169.8, s
8'	86.5, s	86.6, s	86.0, s
9'	50.9, d	51.4, d	51.5, d
10'	45.3, s	45.7, s	45.0, s
11'	121.7, s	121.7, s	128.6, s
12'	173.3, s	173.3, s	173.7, s
13'	55.1, t	54.7, t	54.4, t
14'	24.0, q	24.1, q	24.4, q
15'	73.2, t	73.9, t	69.7, t
1''	166.2, s	171.8, s	167.9, s
2''	114.0, d	129.1, s	129.2, s
3''	152.1, s	136.9, d	137.0, d
4''	66.4, t	61.6, t	12.2, q
5''	14.9, q	14.5, q	14.5, q
6''	172.2, s	172.3, s	
7''	29.0, t	29.3, t	
8''	29.3, t	29.5, t	
9''	172.0, s	171.8, s	

similar to those of **1**, except for the upfield shifts of C-3'' ($\Delta\delta_{\text{C}} -15.2$) and Me-5'' ($\Delta\delta_{\text{H}} -0.38$), as well as the downfield shifts of C-2'' ($\Delta\delta_{\text{C}} 15.1$) and the vinylic proton ($\Delta\delta_{\text{H}} 0.93$) in comparison with compound **1** (Tables 1 and 2). Therefore, Me-5'' was supposed to be located at C-2'' instead of C-3''. This substitution pattern was further supported by the NMR similarity at positions 2'', 3'', and

**Figure 1.** Key ROESY correlations for chloramultilide B (**1**).

5'' of **2** and chloramultilide A. Its relative configuration was also assigned by the ROESY spectrum. Accordingly the structure of **2** was established.

Chloramultilide D (**3**) was obtained as an amorphous powder. The molecular formula $\text{C}_{35}\text{H}_{40}\text{O}_{11}$ was established by the HRES-IMS. The IR spectrum similarity of **3** and **1** indicated a close relationship between these two compounds. Compound **3** resembled most NMR resonances of **1**, except for the absence of those of positions 6'', 7'', 8'', and 9'' (Tables 1 and 2). The lack of this succinyl moiety ($\text{C}_4\text{H}_2\text{O}_3$) in **3** was not only suggested by the NMR data but also reflected by its molecular formula. The chemical shift changes at C-4'' ($\delta_{\text{H}} 1.89$, d, 3H; $\delta_{\text{C}} 12.2$) also supported this conclusion. The low-field chemical shift of the vinylic proton ($\delta_{\text{H}} 6.97$) indicated the location of Me-5'' at C-2'', the same substitution pattern as in **2**. The relative configuration of **3** was also elucidated by the ROESY experiment.

Comparison of NMR and MS data with literature values showed that the known compounds were shizukaol E,⁸ chloramultilide A,¹¹ chlorahololide B (**4**),¹⁵ 8-epiasterolid,¹⁷ hydroxyisogermafureno-

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Supporting Information Available: ^{13}C and ^1H NMR and ROESY spectra for chloramultilides B–D (**1–3**) are available free of charge via the Internet at <http://pubs.acs.org>.

References and Notes

- (1) Jiangsu New Medical College. *Dictionary of Chinese Traditional Medicine*; Shanghai Science and Technology Press: Shanghai, 1990; pp 1759–1760.
- (2) Kawabata, J.; Fukushi, Y.; Tahara, S.; Mizutani, J. *Agric. Biol. Chem.* **1985**, *49*, 1479–1485.
- (3) Kawabata, J.; Mizutani, J. *Agric. Biol. Chem.* **1989**, *53*, 203–207.
- (4) Wu, B.; He, S.; Wu, X.-D.; Pan, Y.-J. *Tetrahedron Lett.* **2007**, *48*, 453–456.
- (5) Kawabata, J.; Fukushi, Y.; Tahara, S.; Mizutani, J. *Phytochemistry* **1990**, *29*, 2332–2334.
- (6) Kawabata, J.; Mizutani, J. *Phytochemistry* **1992**, *31*, 1293–1296.
- (7) Kawabata, J.; Fukushi, E.; Mizutani, J. *Phytochemistry* **1993**, *32*, 1347–1349.
- (8) Kawabata, J.; Fukushi, E.; Mizutani, J. *Phytochemistry* **1995**, *39*, 121–125.
- (9) Kawabata, J.; Fukushi, E.; Mizutani, J. *Phytochemistry* **1998**, *47*, 231–235.
- (10) Takeda, Y.; Yamashita, H.; Matsumoto, T.; Terao, H. *Phytochemistry* **1993**, *33*, 713–715.
- (11) Yang, S.-P.; Yue, J.-M. *Tetrahedron Lett.* **2006**, *47*, 1129–1132.
- (12) Wu, B.; He, S.; Wu, X.-D.; Pan, Y.-J. *Planta Med.* **2006**, *72*, 1334–1338.
- (13) Kawabata, J.; Tahara, S.; Mizutani, J. *Agric. Biol. Chem.* **1981**, *45*, 1447–1453.
- (14) Kwon, O. E.; Lee, H. S.; Lee, S. W.; KiHwan, B.; Koanhoi, K.; Masahiko, H.; Rho, M.-C.; Kim, Y.-K. *J. Ethnopharm.* **2006**, *104*, 270–277.
- (15) Yang, S.-P.; Gao, Z.-B.; Wang, F.-D.; Liao, S.-G.; Chen, H.-D.; Zhang, C.-R.; Hu, G.-Y.; Yue, J.-M. *Org. Lett.* **2007**, *9*, 903–906.
- (16) Spek, A. L. *J. Appl. Crystallogr.* **2003**, *36*, 7–13.
- (17) Bohlmann, F.; Dutta, L. N.; Knauf, W.; Robinson, H.; King, R. M. *Phytochemistry* **1980**, *19*, 433–436.
- (18) Takeda, K.; Horibe, I.; Minato, H. *J. Chem. Soc.* **1968**, 569–572.
- (19) Okamura, H.; Nakashima, N.; Iwagawa, T.; Nakayama, N.; Nakatani, M. *Chem. Lett.* **1994**, 1541–1542.
- (20) Li, Y.; Zhang, D.-M.; Li, J.-B.; Yu, S.-S.; Li, Y.; Li, Y.-M. *J. Nat. Prod.* **2006**, *69*, 616–620.
- (21) National Committee for Clinical Laboratory Standards. Reference method for broth dilution antifungal susceptibility testing of yeasts, Approved standard. NCCL S document M-27-A [J]. Wayne, PA: NCCL S. 1997; pp 1–21.

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